Surface alloying is a powerful way of applying physical and chemical properties of metals, for a number of applications from catalysis to nuclear and green technologies. Surfaces offer many degrees of freedom, giving rise to new phases that do not have a bulk counterpart. However, the atomic characterization of distinct surface compounds is a major task, which demands powerful experimental and theoretical tools. Here we illustrate the process for the case of a GdAu$_2$ surface phase of extraordinary crystallinity. The combined use of surface-sensitive techniques and state-of-the-art ab-initio calculations disentangles its atomic and electronic properties. In particular, the stacking of the surface layers allows for gadolinium’s natural ferromagnetic state, at variance with the bulk phase, where frustration leads to antiferromagnetic inter-layer coupling.

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Many of the greenest technologies of the age, from electric cars to efficient light bulbs to very large wind turbines and future magnetic refrigeration, are made possible by an unusual group of elements: the rare earths (RE). The rare earths possess exotic electronic structure and delicately changing bonding, due to the presence of shallow and interacting d- and f-electron levels. These electrons are responsible for high and specific catalytic activities[1], remarkable alloying in metals[2], and specific luminescent properties[3]. Precisely this richness, and the associated complexity, makes the determination of the structure and properties of RE materials very challenging. The RE demonstrate varying magnetism, coordination, oxidation states, and bonding types, as a function of the number and localization of the f-band electrons. A striking example is Cerium, whose isostructural volume-collapse transition is still not fully explained[4, 5]. All of these properties correspond to and explain the chemical reactivity, tunable optical response, and alloying behavior of the RE and their compounds. The past decade has seen a further revival of RE physics and chemistry, thanks to their interaction with nanostructures, giving rise to novel phases that do not have a bulk counterpart. However, the stacking of the surface layers allows for gadolinium’s natural ferromagnetic state, at variance with the bulk phase, where frustration leads to antiferromagnetic inter-layer coupling.

Reactive growth of Gd on Au(111) is carried out by exposing the Au(111) surface, held at 550 K, to a pure Gd vapor. This procedure leads to alloyed Au-Gd surface structures, that depend on the Gd coverage fraction[12, 13]. In the sub-monolayer range one can observe the smooth transformation of the Au-herringbone reconstruction into a network of trigons, followed by homogeneous nucleation of one monolayer (ML) thick GdAu$_2$ islands. The latter show atomic perfection and a particularly attractive moiré pattern, which arises due to the lattice mismatch with the underlying Au substrate[12, 13]. A second layer grows in a layer-by-layer fashion, as demonstrated in this work. The ARPES data of Figs. 1 (a)-(c) show electronic bands for a Au(111) sample covered by GdAu$_2$ islands one layer thick (Fig. 1(a)), by a full GdAu$_2$ monolayer (Fig. 1(b)), and finally
by a GdAu₂ bilayer (Fig. 1(c)). A sharp transition from 1 ML to 2 ML is observed, reflecting the layered structure of the GdAu₂ surface compound. The topmost surface layer geometry was probed with STM, and corresponds to an expanded GdAu₂(110) plane[12, 13]. However, the bilayer packing is different from that of bulk GdAu₂, as depicted in Figs. 1 (d) and (e). The GdAu₂/Au(111) bilayer structure has been determined with diffraction data and theoretical calculations. The changing distance between the Gd atoms induces a change in the sign of the spin coupling (from AFM in the bulk to FM in the surface alloy), following an RKKY-type interaction[14].

Surface bands in Fig. 1 are measured around the Γ point in the first surface Brillouin zone by ARPES, using 44 eV photons at the Synchrotron radiation Center (SRC) in Stoughton, Wisconsin (see Supplementary Material (SM)). Measurements are carried out at 150 K. The partially covered substrate (Fig. 1(a)) is characterized by the presence of both the Au surface state (SS), whose position is unchanged with respect to that of clean Au(111), and the 1 ML-thick GdAu₂ compound bands. The Au-surface state disappears when the surface is fully covered in Fig. 1(b), and only the GdAu₂ bands remain. One can discriminate three bands, namely A, dispersing upwards from ΓK (better observed at 33.2 eV in Fig. 2), and B and C, dispersing downwards. When the Gd content is further increased, three new bands (D, E, F) are added, closer to E_F. The 2:1 stoichiometry in Figs. 1 (b) and (c) is confirmed by Au and Gd 4f intensities from pure Au(111) and a thick layer of GdAu₂[13], but mainly by the STM tip-surface distance contrast shown in Fig 4 and discussed below. We conclude that bands shown in Figs. 1(b) and (c) correspond to one and two layers of GdAu₂.

The nature of the bands with ARPES can be appropriately determined with first principles calculations, which additionally allows us to test the particular layer stacking in the 2 ML film (we use Density Functional Theory (DFT) - see SM). A (2 × 2) supercell of the Au (111) surface is constructed, on which the different overlayers are deposited. The optimized atomic structure resulting from our calculations, for the double layer compound, consists of two GdAu₂ layers shifted one with respect to the other in such a way that to each Gd atom in one layer corresponds a Au atom in the other layer (Fig. 1 (d)). This packing differs from the bulk termination as seen in Figs. 1 (d-e). The relaxed surface layers show a total surface corrugation of 0.34 Å (distinct from the moiré pattern). Atomic heights h = z_atom − z_Gd are given relative to the sublayer Gd. Three inequivalent atoms coexist in the surface: (1) Au atoms (Au₁) placed above Gd, pushed outwards to h=2.74 Å , (2) Gd atoms sitting on top of Au atoms, which are a bit lower (h=2.44 Å), and (3) Au atoms (Au₂) on top of another Au, with h=2.40 Å. The presence of three inequivalent surface atoms is further proven in STM below. On the other hand, all structures are tested in ferro- (FM) and anti-ferromagnetic (AFM) configurations for the f-electron densities on the Gadolinium. In the bilayer structures the ferromagnetic configuration is preferred in all cases. This can be linked to the different Au-Gd layer stacking. The Au coordination decreases from 10 to 7 or 8 (six in plane plus one or two vertical neighbors). The Gd-Gd distance decreases from 5.2 to 4.12 Å, and the AFM coupling becomes FM (bulk Gd is FM).

In Fig. 2 we compare the theoretical and ARPES electronic structures of the 1ML and 2ML alloys. To estimate the visibility of each electronic band in ARPES, the m=0 angular momentum character of the bands (for all l) is shown as the band width in Fig. 2. The correspondence between theoretical and ARPES bands is quite good, given a possible small shift in the Fermi level due to the finite size of the slab system, the imposed commensurability of the alloy and Au layers (which strains the alloy) and many-body effects which should increase the band widths slightly. Of particular note in the 2ML case are the doubling of the C band, and the flatness of the D band just below E_F. The assignment of the bands to atomic character is straightforward: the manifold below 2 eV (bands B and G) is the Au-d band. The flat band labeled D is of Gd-d character - the corresponding bands in the 1ML case are unoccupied due to charge transfer from Gd to Au. Finally, the surface bands (A, E, F) are hybridized Au-s and Gd-d states, with con-

![FIG. 1: (a-c) Second derivative dispersion plots of the photoemission intensities corresponding to different Gd-Au sample preparations: (a) 0.5 ML GdAu₂/Au(111), (b) 1 ML GdAu₂/Au(111) and (c) 2 ML GdAu₂/Au(111). The ΓK direction refers to the first Brillouin Zone of the GdAu₂ surface (further details in SM). (d) top view of the GdAu₂ bilayer, as determined in DFT. (e) Same for two (110) layers of bulk tetragonal GdAu₂.](image-url)
tributions from both GdAu$_2$ layers in the 2ML case. In particular the upwards dispersing s-like A band appears as a succession of anticrossings in the calculation.

The particular stacking of the bilayer GdAu$_2$ compound is confirmed by x-ray photoelectron diffraction (XPD) experiments (see also SM). The Gd 3$d_{5/2}$ XPD pattern, measured with Si-Kα at a kinetic energy of $E_{\text{kin}}=553$ eV, is shown in Fig. 3(a). To a first approximation this pattern can be interpreted as a forward projected image of the atomic surrounding of photoemitting Gd atoms in the compound. The experimental pattern is dominated by seven intense forward scattering peaks (Fig. 3(a)), which are a clear evidence for the existence of a second GdAu$_2$ layer. The central peak ([111] direction) demonstrates the on-top stacking shown in Fig. 1(d), and excludes the natural bulk stacking (Fig. 1(e)). The other six forward-scattering maxima spaced by 60° in azimuth, at a polar angle of $56° \pm 0.5°$ are due to forward scattering of photoelectrons emitted by Gd atoms below the surface layer. Considering that nearest neighbors are separated by an in-plane distance of $5.5 \pm 0.3$ Å, Gd emitters must be $\sim 2.2$ Å below the surface atoms. The theoretical relaxations give alloy layers with each Gd paired to a Au and not a Gd in the other layer (Fig. 1 - this can not be easily distinguished from XPD). The 2.2 Å agree well with the theoretical distance of $\sim 2.4$ Å. Finally, the Gd 3$d_{5/2}$ XPD pattern is simulated with the Electron Diffraction of Atomic Clusters (EDAC) Multiple Scattering Calculation (MSC) software[15]. The relaxed ab-initio positions are used (in the case of commensurability with the substrate), and the two possible 60° rotated orientations of the surface layer are considered. The final pattern, that best fits the XPD data (Fig. 3(b)), results from the sum of four MSC calculations for Gd atoms in the top and bottom GdAu$_2$ layers when the top layer is 0° or 60° degrees rotated to the bottom one as schematically pictured in (c).

Final proof of the bilayer stacking and stoichiometry comes from the singular tip-height dependence observed in STM images. By tuning the tunneling current ($I_T$), and therefore the tip-sample distance, we are able to image the three inequivalent atoms on the surface predicted by the theoretical calculation. Increasing the current (Fig. 4 a-c), the images go from a hexagonal lattice of holes, to one of protrusions, and to a mixed case, with one intense and one weak protrusion. Such an effect is explainable by the STM tip selectively detecting different electronic states at different heights. The STM images are calculated ab initio using the Tersoff-Hamann scheme [16], and explain the differences observed for a fixed surface topology and bias. For a voltage of $V=-1.2$V, the
STM current density is plotted on the right of Fig. 4 for different values of the partial density of states. As a function of current, the tip lies at different distances above the surface, from 5.19 down to 2.23 Å. At large distances, the Gd and Au$_2$ atoms are visible; in an intermediate range only the Au$_1$ atoms appear, and close to the surface the two Au atoms (inequivalent, due to the sublayer) are visible with different intensities. This inversion of contrasts in STM images at constant bias is rare, but can be understood in light of the bilayer band structure of Fig. 2: at $V=-1.2$ eV bias one is actually probing all electronic states spanning from -1.2 eV up to the Fermi energy. This range comprises the complex hybrid s-$d$ levels, which have different extents at the Au and Gd sites.

In summary, we have shown that the unambiguous determination of the electronic and structural properties of surface phases in alloys requires the combination of cutting-edge surface-sensitive experimental techniques with first-principles modeling. The new stacking sequence found for GdAu$_2$ allows gadolinium’s natural ferromagnetic state, in contrast to bulk GdAu$_2$. Our results would be relevant to explaining the properties of other f-metal complexes.

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