

## The Dissociative Photoionization of O<sub>2</sub> and N<sub>2</sub>O

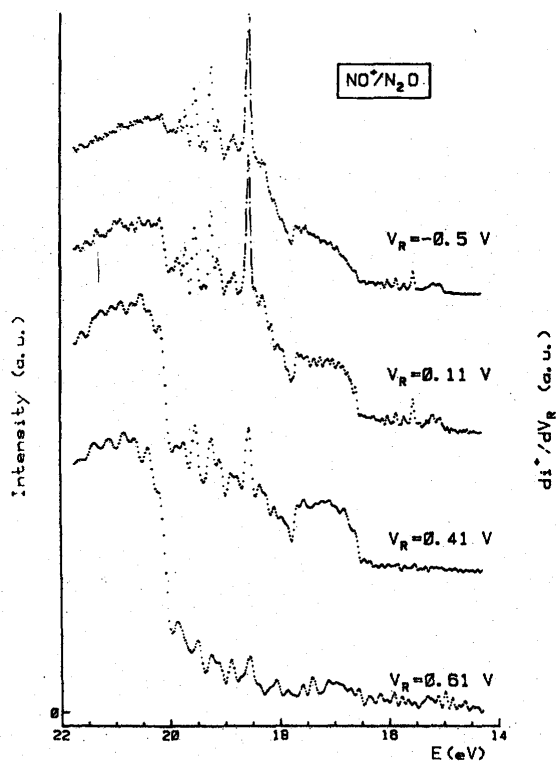
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As dissociative electroionization coupled with ion translational energy analysis gives satisfactory results /1/, the same technique applied to photoionization mass spectrometry has to bring more details (resolution of vibrational energy, autoionization,...).

The method has first been tested on the dissociative photoionization of CH<sub>3</sub>F /2/. Good agreement is found between the electroionization and photoionization results. To test the technique on more complex systems, the photoionization of O<sub>2</sub> and N<sub>2</sub>O has been chosen. In both systems, beside direct dissociative ionization, an abundant autoionization structure is present. Though abundantly investigated /3/, the O<sup>+</sup>/O<sub>2</sub> and N<sub>2</sub>O and NO<sup>+</sup>/N<sub>2</sub>O has been reexamined by photoionization mass spectrometry together with ion energy analysis (retarding potential method). Mainly three aspects are studied. (i) The ion translational energy distributions are examined as a function of the excitation energy. (ii) The photoionization

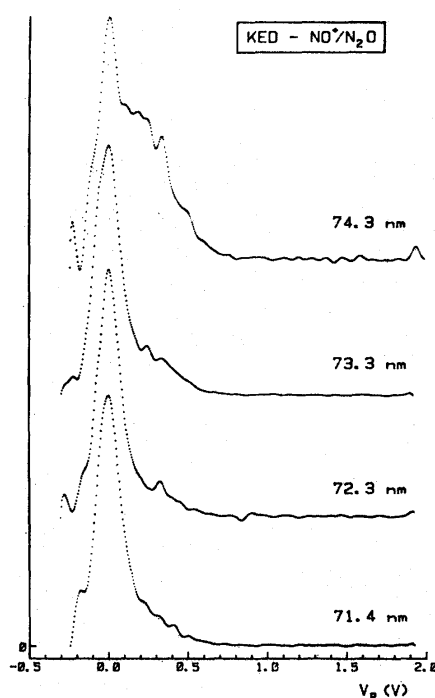
**Fig.1.** PIE curves of NO<sup>+</sup>/N<sub>2</sub>O at V<sub>R</sub> (a)-0.5 V, (b) 0.1 V, and (c) 0.4 V



efficiency curves are recorded as a function of the applied retarding field; a KE-vs-AE plot could be obtained, (iii) Important modifications of the photoelectron spectra of the corresponding molecules, as a function of the excitation energy, are expected. These modifications and those observed in the ion energy distribution spectra could be correlated /4/.

Fig.1 shows a sample of  $\text{NO}^+/\text{N}_2\text{O}$  photoionization efficiency curves at different retarding potential settings. Drastic changes are observed. Fig.2 displays  $\text{NO}^+/\text{N}_2\text{O}$  ion translational energy spectra at different wavelength, i.e. 74.3 nm, 73.3 nm and 72.3 nm corresponding to autoionization processes. The same kind of results are obtained for the  $\text{O}^+$  formation from  $\text{O}_2$  and  $\text{N}_2\text{O}$ .

**Fig.2.** KE spectra of  $\text{NO}^+/\text{N}_2\text{O}$  at (a)74.3 nm,(b)73.3 nm and at (c)72.3 nm.



## References

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