



Fe-Ti-V-P deposits in anorthosite complexes: the bearing of parental magma composition and crystallization conditions on the economic value.

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Recent experimental data (Fram & Longhi 1992; Longhi et al. 1993; Vander Auwera & Longhi 1994; Vander Auwera et al. 1998; Longhi et al. 1999) indicate that parental magmas of the anorthosite-mangerite-charnockite (AMC) suite probably encompass a large continuum of compositions ranging from high-Al basalts (HLCA, Table 1) to more ferroan and potassic compositions, represented by the primitive jotunites (hypersthene-bearing monzodiorites)(TJ, Table 1).

Table 1 Endmember composition of the norite series

	TJ	HLCA
	Vander Auwera & Longhi, CMP 1994	Fram & Longhi, Am Min 1992
SiO ₂	49.39	50.02
TiO ₂	3.67	1.85
Al ₂ O ₃	15.81	17.51
FeO	13.11	10.97
MgO	4.54	6.67
MnO	0.13	0.15
CaO	6.87	8.78
K ₂ O	0.96	0.44
Na ₂ O	3.50	2.93
P ₂ O ₅	0.71	0.16

Experimental phase equilibria show that both endmember magmas can account for the norite series which fractionates at 3-5 kb to silica-enriched liquids (Longhi et al. 1999).

In Rogaland (Fig. 1), comparison between phases experimentally obtained on TJ primitive jotunite (i.e. plag An₄₇ and orthopyroxene En₆₆, Vander Auwera & Longhi 1994) and natural phases from the anorthosite massifs (i.e. An₄₉ and En₇₄ in the Åna-Sira anorthosite massif and An₅₇ and En₇₅ for the Egersund-Ogna anorthosite massif) and from the Bjerkreim-Sokndal layered intrusion (An₅₂ and En₇₇) suggests that a liquid generally similar to this primitive jotunite was parental to both types of intrusions.

Interestingly, Fe-Ti-V-P deposits have been recognized in massive anorthosites and in the Bjerkreim-Sogndal layered intrusion (Duchesne 1999). In massive anorthosites, the ore-bodies occur as (deformed) dykes or pods ranging in composition from pure hemo-ilmenite (Jerneld) to ilmenite norite (Tellnes, Storgangen). Polybaric fractional crystallization and synemplacement deformation in rising anorthosite diapirs lead to relatively Mg- and Cr-rich ilmenite (\pm V-magnetite) deposits. The high contents in Cr and Mg are deleterious for the new chlorination process that tends to substitute to the classical sulfatation process used in the TiO₂ pigment industry. On the other hand fractional crystallization of jotunite magmas in layered magma chambers, such as the Bjerkreim-Sokndal intrusion, gives rise to voluminous "disseminated" mineralizations, containing low Mg and Cr ilmenite + Ti-magnetite \pm REE-rich apatite, more adequate to the chlorination process but still of sub-economic value. Immiscibility is not the controlling mechanism, except maybe in some rare nelsonites (Hesnes).

Subsolidus re-equilibration leads to a thorough change in the oxide mineral composition towards an

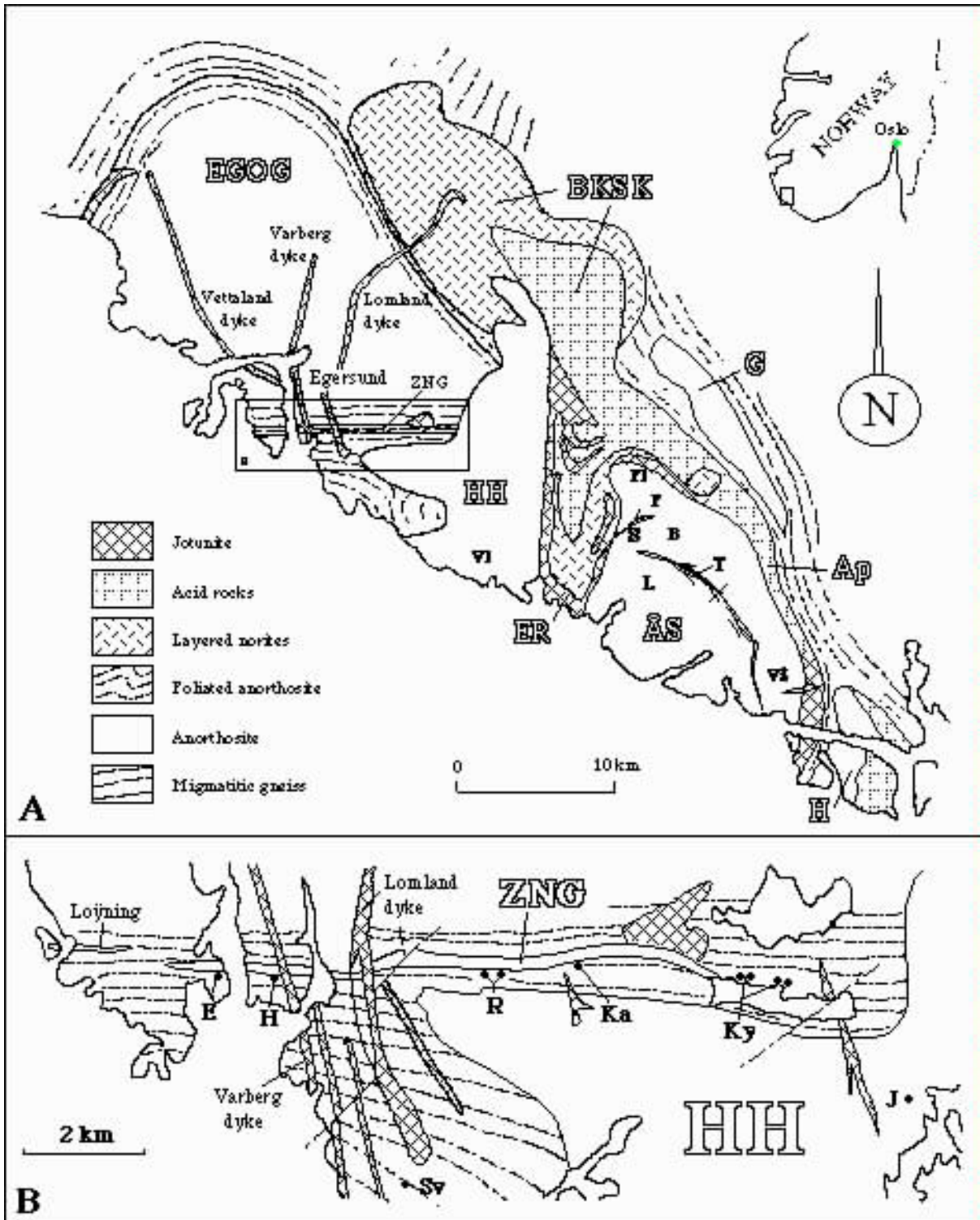


Figure 1 (after Duchesne and Schiellerup, 2001).

A. Schematic geological map of the Rogaland anorthosite province Main geological units: EGOG, Egersund-Ogna massif; HH: Håland-Helleren massif; ÅS: Åna-Sira massif; H: Hidra massif; BSKK: Bjerkreim-Sokndal intrusion; Ap: Apophysis; G: Garsaknat massif; ER: Eia-Rekefjord intrusion. Fe-Ti deposits: B: Blåfjell; F: Flordalen;

Fl: Frøtlog; L: Laksedal; S: Storgangen; T: Tellnes; VI: Vatland.

B. Schematic geological map of the contact zone between the Egersund-Ogna and Håland-Helleren massifs. Same shading as A. ZNG: Norito-granitic Zone; Fe-Ti deposits: E: Eigerøy; H: Hesnes; J: Jerneld; Ka: Kagnuden; Ky: Kydlansvatn; R: Rødemyr; Sv: Svånes

enrichment in end-member compositions. Reactions between the oxide minerals leave conspicuous microscopical evidence : zoning of the hematite exsolution content in the ilmenite grain towards

the contact with a (Ti-) magnetite grain and development of a subcontinuous rim of Al spinel-bearing ilmenite inside magnetite at the contact.. The reactions regularly lower the hematite content of the ilmenite solid solution and the Ti- and Al-spinel contents of the magnetite solid solution. The Mg content of ilmenite also decreases subsolidus, particularly in the Bjerkreim-Sokndal intrusion (Duchesne and Schiellerup, 2001), possibly through reactions with pyroxenes, but no reaction rims can be observed.

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