

Anhydrous Milk Fat enrichment with ¹³C-triacylglycerol tracers: effects on thermal and structural behavior.



Danthine S.¹, Vors C.², Blecker C.¹, Michalski M-C.³



¹Laboratory of Food Science and Formulation , University of Liege, Gembloux Agro Bio-Tech, Gembloux, Belgium

² INSERM U1060, Univ. Lyon-1, CarMeN lab., CRNH-RA, CENS, Oullins, France ³INRA UMR1397, INSA-Lyon, CarMeN lab., Villeurbanne, France

Introduction

The crystallization, melting behavior and polymorphic stability of fats are determined by the behavior of the triacylglycerols (TAGs) they contain. In clinical studies, there is a need to add some ¹³C TAGs as tracers to the ingested fats in order to track their metabolic fate. This procedure could modify physicochemical properties of the fat. The present study was conducted in the framework of a clinical trial aiming at highlighting the effect of the physical structure of a fat (droplets in O/W emulsion or bulk) in a meal on the absorption, chylomicron transport and further metabolic handling of dietary fatty acids (1, 2). We therefore monitored the thermal and polymorphic behavior of anhydrous milk fat (AMF) enriched in tracers (mixture of PPP, OOO and CCC; 1.5 or 5.7 wt%) using DSC and XRD and further compared it to the native AMF.

Material & Methods

DSC melting profiles were recorded using a Q1000 DSC (TA Instruments, USA) according to different procedures:

1) according to AOCS Cj1-94 official method (cooling rate -10°C/min; heating rate 5°C/min)

- 2) Using a tempering procedure in order to mimic the real conditions of the clinical trials:
- Tempering 1 : slow cooling (-1°C/min) to 4°C followed by 15 h tempering at 4°C and subsequent melting from 4 to 70°C (heating rate 15°C/min).
- Tempering 2 : slow cooling (-1°C/min) to 4°C followed by 15 h tempering at 4°C, heating from 4 to 37°C (heating rate 15°C/min), isothermal period of 4h at 37°C, melting from 37 to 70°C (heating rate 15°C/min).

Polymorphism by XRD using a Bruker D8-Advance Diffractometer (Bruker, Germany) and an Anton Paar TTK450 chamber: short and long spacings, after cooling and tempering just as for DSC experiments.

Results : effect of addition of ¹³C TAGs to AMF

DSC Melting profiles and polymorphic stability of Bulk AMF



Native AMF as well as AMF enriched in the low ¹³C TAGs concentration are completely melted at around 37°C, i.e. close to the body temperature.

However, the AMF enriched in high ¹³C TAGs concentration remains crystallized at 37°C. Regarding polymorphic stability, at 4°C, after slow cooling only β ' form is observed in the native AMF Besides β ', the β -form is also detected in the AMF containing high ¹³C TAGs concentration. The β -form even increases during the 4h at 37°C in the AMF sample enriched with high ¹³C TAGs concentration. XRD pattern from 4°C to 37°C, followed by the isothermal period of 4hours at 37°C.



AMF+tracers

from 4°C to

DSC Melting profiles and polymorphic stability of AMF in bulk and emulsified states (low ¹³C TAGs concentration)





Emulsified

Conclusions

Addition of ¹³C TAGs modified AMF melting profile, especially at high concentration. The enriched AMF was completely melted at around 37°C, i.e. close to body temperature. However, under some conditions, AMF enriched in high ¹³C TAGs concentration remained crystallized at 37°C. Similar trends were observed in both systems (bulk vs emulsified). Moreover, AMF polymorphic behavior was also modified upon tracers addition. While only β ' form was observed in native AMF, β form was detected in AMF containing high ¹³C TAGs concentration. Low concentration of tracers should not have any high impact on human digestive physiology. However more attention should be paid to physicochemical structure when high concentrations are added.

(1) Vors et al. 2013. Modulating absorption and postprandial handling of dietary fatty acids by structuring fat in the meal: a randomized cross-over clinical trial. Am J Clin Nutr, 97(1): 23-36. (2) Gabert et al. 2011. ¹³C tracer recovery in human stools after digestion of a fat-rich meal labelled with [1,1,1-¹³C3] tripalmitin and [1,1,1-¹³C3 25(19):2697-703 sabine.danthine@ulg.ac.be