





# Influence of whey protein denaturation on adherence of soiling particles to stainless steel

Y. Touré, P.G. Rouxhet, C.C. Dupont-Gillain and M. Sindic

yetioman.toure@doct.ulg.ac.be

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# **Presentation outline**

# **1. Introduction**

Background

Objectives

# 2. Experimental aspects

**Material** 

Methods

# 3. Results and discussion

4. Conclusion

### Background

#### Concern ? Particulate deposition on surface - drying →natural environments →industrial equipments

#### Where ?

- Storage tanks
- Ducts
- Plates of heaters coolers

→Food processing

### Background

#### Soil attachment and removal: influence of macromolecules at particule-substrate interface

Detry et al. (2011): starch deposit

- presence of macromolecules (polysaccarides, proteins)
- accumulation at substrateparticulate interface
- influence of:
  - details in the mode of drying
  - exposure to moisture



20 µm

# Background

Touré et al et al. (2011; 2013): quartz particles deposit

> dextran, weak effect, low adsorption easy desorption



# Background

Touré et al et al. (2011; 2013): quartz particles deposit

- BSA:
- on polystyrene: negligible effect
- on glass: drastic 7 cleanability prevention of tight bonds induction of a repulsion



Background

#### Protein used in food industry improves

### fabricated foods qualities

- texture
- appearance

#### Denaturation of whey protein and fouling:

- deposit build-up and removal controversial question
- particle soil adherence no data available

### Background

#### Understanding the influence of protein denaturation on particulate soils adherence and removal

- practical information on
  - incidence of surface properties of soil and substrate
  - influence of biomacromolecules
- physico-chemical mechanisms involved :
  - interactions solid-solution, solid-solid
  - biomacrolecules at interfaces
- designing easy-to-clean surfaces

#### Understanding the influence of protein denaturation on particulate soils adherence and removal

This work:

to improve

- evaluating cleanability
- understanding mechanisms involved

Material

Model of whey protein :  $\beta$ -lactoglobulin ( $\beta$ -LGB)

Model of substrate : stainless steel

Model of particulate soil : suspension of quartz particles (10-30 µm)

Substrate pretreatment



### Soil preparation



# Soiling procedure



# **Cleaning test**









### • At given flow rate:

 larger critical detachment radius
←→ lower hydrodynamic drag force required to detach particles





### • At given flow rate:

 larger critical detachment radius
←→ lower hydrodynamic drag force required to detach particles

 Increasing the flow rate same critical detachment radius
←→ increasing drag force required to detach particles





# **Characterization**

#### **Solution and supernatants**

- UV-visible absorption
- soluble protein concentration
- liquid surface tension

Substrate

- static contact angle
  - water
  - supernatants
- XPS analysis
  - bare substrate
  - conditioned substrates



# Solution and supernatants



	Native	Supernatant 0.5 h	Supernatant 4 h
Concentration (g/L)	$2.5\pm0.2$	$\textbf{1.1}\pm\textbf{0.2}$	$\textbf{0.6}\pm\textbf{0.1}$

### Heating β-LGB solution at 75°C

- 0.5 h  $\rightarrow$  aggregation 50%
- 4 h  $\rightarrow$  aggregation 75%,
  - lower surface tension
    - ➔ higher activity at water/air interface

### Surface chemical composition

Conditioning			Proportion (	%) of elements due to	Molar ratios in organic adlayer		
liquid	Heating	Rinsing	adlayer	substrate	O <sub>org</sub> /C <sub>tot</sub>	N/C <sub>tot</sub>	O <sub>org</sub> /N
none	none 4 h						
-β-LGB	none	none					
		twice					
	0.5 h	none					
		twice					
	4 h	none					
		twice					
supernatant 0.5 h	none	none					
		twice					
supernatant 4 h	none	none					
		twice					
computed for β-LGB							

### Computation

 $O_{org}$  and  $O_{inorg}$  from XPS spectra adlayer =  $C_{tot} + N + O_{org}$ substrate = Fe + Cr +  $O_{inorg}$ 

# Surface chemical composition

Conditioning			Proportion (%) of elements due to Molar ratios in organic adlaye				nic adlayer
liquid	Heating	Rinsing	adlayer	substrate	O <sub>org</sub> /C <sub>tot</sub>	N/C <sub>tot</sub>	O <sub>org</sub> /N
none	none		55	45			
	4 h		58	42			
-β-LGB	none	none	96	4			
		twice	81	19			
	0.5 h	none	96	4			
		twice	78	22			
	4 h	none	101	-1			
		twice	103	-3			
supernatant 0.5 h	none	none	98	2			
		twice	94	6			
supernatant 4 h	none	none	82	18			
		twice	77	23			
computed for β-LGB		100	0				

#### **Bare samples**

presence of organic contaminants

### **Conditioned samples**

- organic layer: dominated by protein
- rinsing : weak desorption, same as BSA and others substrates
  - $\rightarrow$   $\beta$ -LGB, not quickly desorbed during cleaning test

# **Contact angle**



#### Water contact angle

higher than expected for surface consisting of chromium and iron oxides: presence of organic contaminants

#### Supernatant 4 h





### **Critical detachment radius**



#### Presence of $\beta$ -LGB :

- native : decrease of critical detachment radius
  - = higher adherence
- denatured (heated 4 h) : further increase of adherence

### **Critical detachment radius**





Heating after soiling with

suspension in water

higher adherence

↔ capillary forces

• suspension in native β-LGB

still higher adherence

### **Critical detachment radius**



#### Presence of β-LGB

- slightly increased adherence
- contradiction with

BSA/glass, β-LGB/glass, β-LGB/stainless steel cleaned UVO

due to surface contamination with organic compounds

#### Presence of denatured β-LGB

further increased adherence of soiling particles

### 4. Conclusion

**Presence of**  $\beta$ **-LGB**:

- in soiling quartz particles suspension,
  - adsorption and domination of in organic layer
  - limited desorption upon rinsing
- at the quartz particle/stainless steel interface, particle adherence **7**, enhanced by denaturation

### Stainless steel vs systems investigated before:

protein influence via

- droplet spreading
- soiling particles aggregation

→

minor importance with respect to

direct effects at interface

Remark on influence of stainless steel surface state stainless steel does not behave as a hydrophilic substrate owing to surface contamination with organic compounds

### Broader study, including substrate hydrophobicity is under way

Thank you for your attention! Are there any questions?