

## CO<sub>2</sub> dynamics and related air-ice-sea gas transfer in spring pack and land fast sea ice

Delille B.<sup>1\*</sup>, V. Schoemann<sup>2</sup>, C. Lancelot<sup>2</sup>, D. Lannuzel<sup>3</sup>, J.T.M. de Jong<sup>3</sup>, B. Tilbrook<sup>4,5</sup>,  
D. Delille<sup>6</sup>, A.V. Borges<sup>1</sup>, J.-L. Tison<sup>7</sup>

<sup>1</sup> *Chemical Oceanography Unit, Université de Liège, Belgium*

<sup>2</sup> *Ecologie des Systèmes Aquatiques, Université Libre de Bruxelles, Belgium*

<sup>3</sup> *Océanographie Chimique et Géochimie des Eaux, Université Libre de Bruxelles, Belgium*

<sup>4</sup> *CSIRO Marine Research, Australia*

<sup>5</sup> *ACE CRC and Antarctic Division, University of Tasmania, Australia*

<sup>6</sup> *Observatoire Océanologique de Banyuls, Université P. et M. Curie, France*

<sup>7</sup> *Glaciology Unit, Université Libre de Bruxelles, Belgium*

There are growing observations that sea ice exchange CO<sub>2</sub> directly with the atmosphere. To explore the relationships between sea ice-specific biogeochemical processes and fluxes of CO<sub>2</sub> at the air-ice interface, we carried out three surveys which addressed spring and summer CO<sub>2</sub> dynamics in Antarctic land fast sea ice, and first year and multiyear pack ice.

Spring and summer pCO<sub>2</sub> patterns are consistent between the three surveys and mainly result from physical processes (temperature increase and related melting, convection of brines,...) while the under-saturation observed in summer is the signature of chemical (dissolution of carbonate minerals) and biological processes within sea ice. Exchanges of CO<sub>2</sub> at the air-ice interface are unsurprisingly driven by the evolution of pCO<sub>2</sub> within sea ice, yet modulated by sea ice permeability.

Cold ice is generally not permeable either to gas or water transfer. As temperature crosses the threshold value of about -5°C, sea ice becomes permeable to gas, and sea ice begins to release CO<sub>2</sub> to the atmosphere at a rate up to 1.9 mmol.m<sup>-2</sup>.d<sup>-1</sup>. However, as the ice continues to warm up, pCO<sub>2</sub> decreases dramatically to reach under-saturation of CO<sub>2</sub> (pCO<sub>2</sub> down to 30 ppmV). and sea ice turns into a CO<sub>2</sub> sink with CO<sub>2</sub> fluxes ranging up to -6 mmol.m<sup>-2</sup>.d<sup>-1</sup>. First tentative, and probably underestimated, budgets of air-ice CO<sub>2</sub> fluxes point out that Antarctic sea ice edge would represent an additional CO<sub>2</sub> sink of 6 to 9 % to the current estimate of the uptake of the Southern Ocean south of 50°S. We assessed how realistic could be such CO<sub>2</sub> fluxes by estimating the potential CO<sub>2</sub> fluxes driven by each main sea ice biogeochemical processes. This independent assessment is consistent with estimates derived from air-sea CO<sub>2</sub> fluxes measurements and point out the significance of abiotic sea ice-specific biogeochemical processes.

\* Bruno.Delille@ulg.ac.be